

Chhar Tola to Sugauli

(Hill maintenance policy) Agency: Maa Bandhuvani
Caulh

Measurement Book No-311

Schedule XLV-Form No. 134

Agrib - 03MBD/2019-20

Bagaha - 2 DIVISION

Piprasi SUB-DIVISION

DOS - 27-01-2020

BSC - 26-10-2020

Particulars	Details of actual measurement				Contents of area
	No.	L.	B.	D.	
R.F —————					37187518.w
Below 0.110/o ————	(-)				40906.w
					N.A - 37146612.w
					13705720
					AB
					215/2020
					OE
					<u>material statement</u>
E/W —					3519m ³
stone metal —					2792.67m ³

material	statement
1. The first law of thermodynamics states that energy cannot be created or destroyed.	True
2. The second law of thermodynamics states that the total entropy of an isolated system can never decrease over time.	True
3. The third law of thermodynamics states that the entropy of a perfect crystal at absolute zero is zero.	True
4. The zeroth law of thermodynamics states that if two systems are each in thermal equilibrium with a third system, then they are in thermal equilibrium with each other.	True
5. The first law of thermodynamics states that the total energy of a closed system is constant.	True
6. The second law of thermodynamics states that the entropy of a closed system always increases over time.	True
7. The third law of thermodynamics states that the entropy of a system approaches a constant value as the temperature approaches absolute zero.	True
8. The zeroth law of thermodynamics states that if two systems are in thermal contact and no net energy is exchanged between them, then they are in thermal equilibrium.	True
9. The first law of thermodynamics states that the change in internal energy of a system is equal to the heat added to the system minus the work done by the system.	True
10. The second law of thermodynamics states that the entropy of a system increases when heat is added to the system.	True
11. The third law of thermodynamics states that the entropy of a system is zero at absolute zero.	True
12. The zeroth law of thermodynamics states that if two systems are in thermal equilibrium with each other, then they are in thermal equilibrium with any third system.	True
13. The first law of thermodynamics states that the total energy of a system is constant.	True
14. The second law of thermodynamics states that the entropy of a system always increases over time.	True
15. The third law of thermodynamics states that the entropy of a system approaches a constant value as the temperature approaches absolute zero.	True
16. The zeroth law of thermodynamics states that if two systems are in thermal contact and no net energy is exchanged between them, then they are in thermal equilibrium.	True
17. The first law of thermodynamics states that the change in internal energy of a system is equal to the heat added to the system minus the work done by the system.	True
18. The second law of thermodynamics states that the entropy of a system increases when heat is added to the system.	True
19. The third law of thermodynamics states that the entropy of a system is zero at absolute zero.	True
20. The zeroth law of thermodynamics states that if two systems are in thermal equilibrium with each other, then they are in thermal equilibrium with any third system.	True

A/W -	35/9m	3
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Stomemita - 2792.67 m³

Storm screening - 319.40 m³

Local sand - 289.70 m³

Bitumen (590) — 160656 kg

Emulsion (SS4) - 1166218

Formzahl (RS) - 127/528

stone chips - 2042.99 m³

1 am

125 2020
75

12/05/20 AB